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53. ChemQuest 17. Name: \_\_\_\_\_

Date: \_\_\_\_\_ Hour: \_\_\_\_\_

Information: Separating Charges.  
Examine Figure 1 below where there are three pairs of metal spheres that have different amounts of. charge on them. The spheres in diagram C are closer than the others because they have the strongest. attraction.

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*AP Chemistry 2017-18*

This video is designed to help you complete the molarity chemquest found in Unit 11.

*Molarity*

118 ChemQuest 39 Name: \_\_\_\_\_

Date: \_\_\_\_\_ Hour: \_\_\_\_\_

Information: Molarity

Concentration is a term that describes the amount of solute that is dissolved in a solution.

Concentrated solutions contain a lot of dissolved solute, but dilute solutions contain only a little.

*118 ChemQuest 39 Name:*

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An element exists as 4 different isotopes. 4.35% have a mass of 49.9461, 83.79% have a mass of 51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this

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element? b. What is the identity of this element?

*Chemistry: ChemQuest Skill  
Practice (Semester 1 ...*

traditionally and ... Chemquest 23  
Answer Key More Hint: add your  
answer to question 1 and four  
times your answer to question 2.  
 $4 + 4(7) = 4 + 28 = 32$  4.

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question 1 and four times your  
answer to question 2.  $4 + 4(7) =$   
 $4 + 28 = 32$  4. Chemistry Inquiry

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15. Given your answer to question

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14 and the reaction you wrote in question 13, calculate the molarity of the nitrate ions ( $\text{NO}_3^-$ ) in the solution. Title: Molarity ChemQuest WS.pages Created Date:

## *Molarity ChemQuest WS - MS MCLARTY'S CLASSES*

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of  $\text{CO}_2$  are produced, so the number of moles of  $\text{CO}_2$  that can be produced when 4 moles of propane combusts =  $4(3) = 12$ .

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